



JEE ANSWER BOOKLET

SUBJECTIVE ASSESSMENT - I

Ac. Yr. 2018-2019

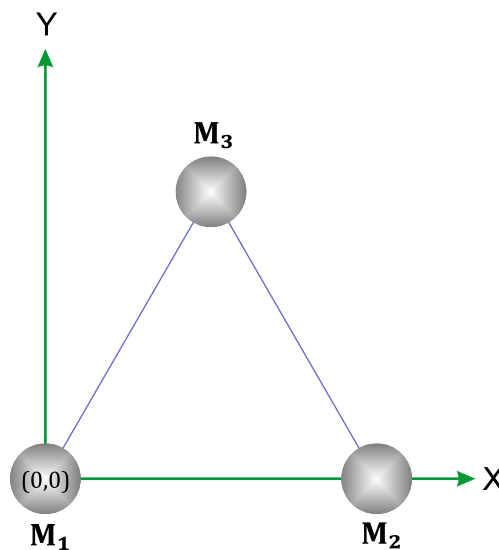
NAME OF STUDENT :

CHAPTERS: **ROTATIONAL MECHANICS**
IONIC EQUILIBRIUM
DIFFERENTIATION AND APPLICATION OF DERIVATIVES

DATE OF ASSESSMENT / /

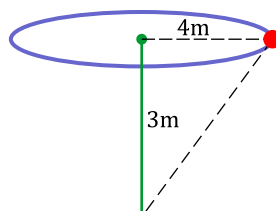
MARKS SCORED

Q.1. (a) Locate the centre of mass of a system of particles of masses $m_1 = 1$ kg, $m_2 = 2$ kg and $m_3 = 3$ kg, situated at the corners of an equilateral triangle of side 1 metre.



Q.1. (b) A particle of mass 10 kg is moving in a circle of 4 m radius with a constant speed of 5 m/sec. What is the angular momentum about:

- (i) the centre of the circle?
- (ii) and 3 m distant from its centre?
- (iii) Which one of these will always be in the same direction?





QUESTIONS TO BE SOLVED ON THIS PAGE: Q.3.

Q.3. On the flat surface of a disc of radius a , a small circular hole of radius b is made with its centre at a distance c from the centre of the disc.

If mass of the whole uncut disc is M ,

Calculate :

(i) the new position of the centre of mass if centre of mass of uncut disc is the origin.

(ii) the moment of inertia of the holed disc about the axis of the circular hole.

ROUGH WORK





QUESTIONS TO BE SOLVED ON THIS PAGE: Q.4.

Q.4. If the tangent at a point (1, 2) on the curve $y = ax^2 + bx + 7/2$ be parallel to the normal at point (-2, 2) on the curve $y = x^2 + 6x + 10$, then find the value of a and b .

Main solution area with horizontal lines for writing the answer.

Main solution area with horizontal lines for writing the answer.

ROUGH WORK

Rough work area with horizontal lines for calculations.



QUESTIONS TO BE SOLVED ON THIS PAGE: Q.6.

Q.6. Find the greatest and least values of the following functions on the indicated intervals:

(i) $f(x) = 2x^3 - 3x^2 - 12x + 1$ on $[-2, 5/2]$

(ii) $f(x) = x^2 \log x$ on $[1, e]$

ROUGH WORK



QUESTIONS TO BE SOLVED ON THIS PAGE: Q.7.

Q.7. The degree of ionization of a 0.1M bromoacetic acid solution is 0.132. Calculate the pH of the solution and the pK_a of bromoacetic acid.

ROUGH WORK

QUESTIONS TO BE SOLVED ON THIS PAGE: Q.8.

Q.8. The concentration of sulphide ion in 0.1M HCl solution saturated with hydrogen sulphide is 1.0×10^{-19} M. If 10 mL of this is added to 5 mL of 0.04 M solution of the following: FeSO₄, MnCl₂, ZnCl₂ and CdCl₂, in which of these solutions precipitation will take place?

ROUGH WORK

QUESTIONS TO BE SOLVED ON THIS PAGE: Q.9.

Q.9. The ionization constant of nitrous acid is 4.5×10^{-4} . Calculate the pH of 0.04 M sodium nitrite solution and also its degree of hydrolysis.

Area for writing the solution to the question. The space contains ten horizontal lines for the student's work.

Area for writing the solution to the question. This section contains ten vertical lines and is partially enclosed by a vertical line on the left and a vertical line on the right.

ROUGH WORK

Area for rough work or calculations. This section is a large empty rectangular box on the right side of the page.

ROUGH WORK

Lined area for rough work.

